

## Practice Problem Sheet II – Fall 2024

1. Given

$$H = \sum_{i=1}^N \frac{p_i^2}{2m},$$

you can write the surface area of the energy shell at  $E$  in a microcanonical (MC) ensemble as

$$\Omega(E) = \frac{1}{h^{3N} N!} \int \int d^3x_1 \dots d^3x_N d^3p_1 \dots d^3p_N \delta \left( E - \sum_{i=1}^N \frac{p_i^2}{2m} \right).$$

Evaluate the integral above in MC ensembles. (Hint: This is an ideal gas problem in MC)

2. A Hamiltonian (for a two-level system) can be written as

$$H = -h \sum_{i=1}^N \sigma_i, \quad \sigma_i = \pm 1.$$

a) Show that in microcanonical ensembles, by writing,

$$\Omega(E) = \sum_{i=1}^N \delta(E + h \sum_{\sigma_i = \pm 1} \sigma_i) = 2^N \int \frac{dk}{2\pi} e^{f(k)}$$

one obtains an expression of  $f(k)$ ,

$$f(k) = ikE + N \ln \cos kh. \quad (i^2 = -1)$$

b) Compute the integral by using the saddle-point approximation and show that

$$S = k_B \ln \Omega(E) = -\frac{k_B N}{2} \left[ (1+x) \ln \frac{1+x}{2} + (1-x) \ln \frac{1-x}{2} \right]$$

where  $x = E/(Nh)$ .

c) Calculate the temperature  $T$  from the definition of entropy and plot  $T$  as a function of  $x$ . Discuss your results for  $x \geq 0$ .

3. Show that in canonical ensembles the specific heat  $C_v$  is given by,

$$C_v = k_B \beta^2 \left[ \langle E^2 \rangle - \langle E \rangle^2 \right].$$

4. The partition function of a system in canonical ensembles is given by

$$Z = \frac{x}{1-x}, \quad \text{where } x = e^{-\frac{\beta h \omega}{2}}.$$

Calculate  $U, C_v, F$  and  $S$  for this system.

5. A system consists of two energy levels  $-\Delta/2$  and  $\Delta/2$ . Write down the the partition function of the system in canonical ensembles and calculate the specific heat  $C_v$